13.4: Mutagens and Carcinogens

A carcinogen is any agent that directly increases the incidence of cancer. Most, but not all carcinogens are mutagens. Carcinogens that do not directly damage DNA include substances that accelerate cell division, thereby leaving less opportunity for cell to repair induced mutations, or errors in replication. Carcinogens that act as mutagens may be biological, physical, or chemical in nature, although the term is most often used in relation to chemical substances.

Human Papilloma Virus (HPV, Figure 13.4) is an example of a biological carcinogen. Almost all cervical cancers begin with infection by HPV, which contains genes that disrupt the normal pattern of cell division within the host cell. Any gene that leads to an uncontrolled increase in cell division is called an oncogene. The HPV E6 and E7 genes are considered oncogenes because they inhibit the host cell's natural tumor suppressing proteins (include p53, described below). The product of the E5 gene mimics the host's own signals for cell division, and these and other viral gene products may contribute to dysplasia, which is detected during a Pap smear (Figure 13.5). Detection of abnormal cell morphology in a Pap smear is not necessarily evidence of cancer. It must be emphasized again that cells have many regulatory mechanisms to limit division and growth, and for cancer to occur, each of these mechanisms must be disrupted. This is one reason why only a minority of individuals with HPV infections ultimately develop cancer. Although most HPV-related cancers are cervical, HPV infection can also lead to cancer in other tissues, in both women and men.
Radiation is a well-known physical carcinogen, because of its potential to induce DNA damage within the body. The most damaging type of radiation is **ionizing**, meaning waves or particles with sufficient energy to strip electrons from the molecules they encounter, including DNA or molecules that can subsequently react with DNA. Ionizing radiation, which includes x-rays, gamma rays, and some wavelengths of ultraviolet rays, is distinct from the non-ionizing radiation of microwave ovens, cell phones, and radios. As with other carcinogens, mutation of multiple, independent genes that normally regulate cell division is required before cancer develops.

Chemical carcinogens (Table 13.2) can be either natural or synthetic compounds that, based on animal feeding trials or **epidemiological** (i.e. human population) studies, increase the incidence of cancer. The definition of a chemical as a carcinogen is problematic for several reasons. Some chemicals become carcinogenic only after they are metabolized into another compound in the body; not all species or individuals may metabolize chemicals in the same way. Also, the carcinogenic properties of a compound are usually dependent on its dose. It can be difficult to define a relevant dose for both lab animals and humans. Nevertheless, when a correlation between cancer incidence and chemical exposure is observed, it is usually possible to find ways to reduce exposure to that chemical.

**Table 13.2: Some classes of chemical carcinogens (Pecorino 2008)**

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Figure 13.4: 
Electron micrograph of HPV.  
(Wikipedia-Unknown-PD)

Figure 13.5:  
Dysplastic (left) and normal (right) cells from a Pap smear.  
(Flickr-Ed Uthman-CC:AS)
1. PAHs (polycyclic aromatic hydrocarbons)
   e.g. benzo[a]pyrene and several other components of the smoke of cigarettes, wood, and fossil fuels

2. Aromatic amines
   e.g. formed in food when meat (including fish, poultry) are cooked at high temperature

3. Nitrosamines and nitrosamides
   e.g. found in tobacco and in some smoked meat and fish

4. Azo dyes
   e.g. various dyes and pigments used in textiles, leather, paints.

5. Carbamates
   e.g. ethyl carbamate (urethane) found in some distilled beverages and fermented foods

6. Halogenated compounds
   e.g. pentachlorophenol used in some wood preservatives and pesticides.

7. Inorganic compounds
   e.g. asbestos; may induce chronic inflammation and reactive oxygen species

8. Miscellaneous compounds
   e.g. alkylating agents, phenolics

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